

# Convert Pressure

$$dp = -\rho g dz$$

$$dp = -\frac{p}{R_s T} g dz$$

$$p = p_0 * \left(\frac{T}{T_0}\right)^{-\frac{g}{R_d \gamma}} \quad \text{constant: } \gamma = \frac{dT}{dz}$$

$$p = p_0 * \left(\frac{T}{T_0}\right)^{-\frac{g}{R_d \gamma}}$$

$$p = p_0 * \left(\frac{T}{T - \gamma * h}\right)^{-\frac{g}{R_d \gamma}} \quad (\because T_0 = T - \gamma * h)$$

$$p = p_0 * \left(1 + \frac{\gamma * h}{T - \gamma * h}\right)^{-\frac{g}{R_d \gamma}}$$

$$\therefore p = p_0 * \left(1 - \frac{0.0065 * h}{T(^{\circ}\text{C}) + 273.15 + 0.0065 * h}\right)^{5.257}, \quad p_0 = p * \left(1 - \frac{0.0065 * h}{T(^{\circ}\text{C}) + 273.15 + 0.0065 * h}\right)^{-5.257}$$

$R_s$	: specific gas constant (J/kg/K)
$R_d$	: specific gas constant of dry air (287 J/kg/K)
$g$	: 9.80665 m/s <sup>2</sup>
$\gamma$	: -0.0065 K/m